**Research article** 

# Hosoya, Schultz, Modified Schultz Polynomials and Their Topological Indices of *Benzene Molecules*: First Members of *Polycyclic Aromatic Hydrocarbons (PAHs)*

#### Mohammad Reza Farahani,

Department of Applied Mathematics,

Iran University of Science and Technology (IUST),

Narmak, Tehran 16844, Iran.

E-mail: MR\_Farahani@mathdep.iust.ac.ir

MRFarahani88@gmail.com

#### Abstract

A topological index is a real number related to the structure of a connected graph *G* and is invariant under graph automorphism. Let *G* be a (molecular) graph possessing *n* vertices and *m* edges, and e=x,y be an edge of *G* and *x*, *y* are two of its vertices, then the distance d(x,y)=d(x,y|G) between the vertices *x* and *y* is equal to the length of the shortest path that connects them in *G*.

The goal of this paper is to compute the Schultz Polynomial  $Sc(G,x) = \frac{1}{2} \sum_{(u,v) \in V(G)} (d_u + d_v) x^{d(u,v)}$ , Modified Schultz Polynomial  $Sc^*(G,x) = \frac{1}{2} \sum_{(u,v) \in V(G)} (d_u d_v) x^{d(u,v)}$ , Hosoya Polynomial  $H(G,x) = \frac{1}{2} \sum_{(u,v) \in V(G)} x^{d(v,u)}$  and their topological indices of first members of a family of hydrocarbon structures "*Polycyclic Aromatic Hydrocarbons (PAHs)*" molecular graph. **Copyright © acascipub.com, all rights reserved.** 

**Keywords:** Polycyclic Aromatic Hydrocarbons (PAHs), Benzenoid, Topological Indices, Wiener Index, Hosoya Polynomial, Schultz Polynomial, Modified Schultz Polynomial.

#### Introduction

Let G=(V,E) a simple finite molecular graph with the vertex set V(G) (the number of vertices |V(G)|=n) and the edge set E(G) (the number of vertices |E(G)|=m), such that in the connected molecular graph G, vertices represent atoms and edges represent bonds.

In graph theory, if *e* is an edge of G, connecting the vertices *u* and *v*, then we write e=uv. If G is a connected graph and *x* and *y* are two of its vertices, then the distance

d(x,y)=d(x,y|G) between the vertices x and y is equal to the length of the shortest path that connects them in G. Also the number of adjacent of vertex v is its degree, which denoted by  $d_v$ .

In chemical graph theory, we have invariant polynomials for any graphs, that they have usually integer coefficients. A topological index of G is a numeric quantity, derived following certain rules in Chemistry, which can be used to characterize the property of molecule.

Usage of topological indices in Biology and Chemistry began in 1947 when chemist *Harold Wiener* [1] introduced Wiener index to demonstrate correlations between physico-chemical properties of organic compounds of molecular graphs. The Wiener number is sum of distances between all unordered pairs of vertices of a simple graph *G*.

Also, for this topological index, the Hosoya polynomial was introduced by *H. Hosoya*, in 1988 [2]. The Wiener index W(G) and its polynomial (Hosoya) H(G,x) are define as follow:

$$W(G) = \frac{1}{2} \sum_{v \in V(G)} \sum_{u \in V(G)} d(v, u)$$
$$H(G, x) = \frac{1}{2} \sum_{v \in V(G)} \sum_{u \in V(G)} x^{d(v, u)}$$

Another based structure descriptors is the "molecular topological index" (Schultz index) was introduced by *Harry P. Schultz* in 1989 [3] and the Modified Schultz index was defined by *S. Klavžar* and *I. Gutman* in 1997 [4].

The Schultz index is defined as:

$$Sc(G) = \frac{1}{2} \sum_{\{u,v\} \subset V(G)} (d_u + d_v) d(u,v)$$

where  $d_u$  and  $d_v$  are degrees of vertices u and v.

And the Modified Schultz polynomial of *G* is defined as:

$$Sc^{*}(G) = \frac{1}{2} \sum_{\{u,v\} \in V(G)} d(u,v) (d_{u}d_{v})$$

Also for these One of the important, we have two important polynomials "*Schultz polynomial*" and "*Modified Schultz polynomial*". Schultz and Modified Schultz polynomials of *G* are defined respectively as:

$$Sc(G,x) = \frac{1}{2} \sum_{\{u,v\} \subset V(G)} (d_u + d_v) x^{d(u,v)}$$

10

International Journal of Theoretical Chemistry

Vol. 1, No. 2, October 2013, PP: 09 - 16

Available online at http://acascipub.com/Journals.php

$$Sc^{*}(G,x) = \frac{1}{2} \sum_{\{u,v\} \subset V(G)} (d_{u}d_{v}) x^{d(u,v)}$$

These based structure descriptors and their polynomials studied and computed in many papers [3-15].

In this paper, we focus on the structure of first member of *Polycyclic Aromatic Hydrocarbons (PAHs)*, which called *Bezene* molecules. And formulas of its Hosoya, Schultz, Modified Schultz polynomials and their topological indices are determined for the first time.

### **Results and Discussion**

In this section we compute Hosoya, Schultz and Modified Schultz polynomials, Wiener, Schultz and Modified Schultz indices of first members of *Polycyclic Aromatic Hydrocarbons* molecules, which called *Benzene PAH*<sub>1</sub> in Theorem1.

*Polycyclic Aromatic Hydrocarbons PAHn* is a family of hydrocarbon molecules, such that its structure is consisting of cycles with length six (*Benzene*).

In Refs [16-29] some properties and more historical details of this family of hydrocarbon molecules are studed. Also polycyclic aromatic hydrocarbons *PAHn* family are very similar properties to one of famous family of *Benzenoid* system (*Circumcoronene Homologous Series of Benzenoid*  $H_k$ ). The properties and applications of Benzenoid system are presented in many papers; reader can see references [30-43].

**Theorem 1.** Let  $PAH_1$  be the first members of *Polycyclic Aromatic Hydrocarbons (PAHs)*: "*Benzene* molecules". Then Hosoya, Schultz and Modified Schultz polynomials of  $PAH_1$  are equal to

- $H(PAH_1, x) = 12x^1 + 18x^2 + 21x^3 + 12x^4 + 3x^5$
- $SC(PAH_1, x) = 60x^1 + 96x^2 + 78x^3 + 36x^4 + 6x^5$
- $SC^*(PAH_1,x) = 72x^1 + 180x^2 + 69x^3 + 24x^{4+}3x^5$

And also the following topological indices of Benzene  $PAH_1$  are calculated by formulas:

- Wiener index :  $W(PAH_1) = 164$
- Schultz index :  $SC(PAH_1)=660$
- Modified Schultz index :  $SC^*(PAH_1) = 750$

Before prove the main theorem, we need the following denotations.

Denotation 1. Let d(u,v)=i is distance between vertices u and v of G. Then,  $D_i = \{(u,v) | u, v \in V(G), d(u,v)=i\}$  and we denoted the size of  $D_i$  by d(G,i). Thus the diameter d(G) is the longest topological distance in G.



Figure 1. First member of of Polycyclic Aromatic Hydrocarbons (PAHs): "Benzene molecules"

**Proof of Theorem 1:** Let  $G=PAH_1$  be the *Benzene* (frist member of the polycyclic aromatic hydrocarbon family) with six carbon (C) and six hydrogen (H) atoms. Obviously we have the  $\binom{12}{2}=66$  distinct shortest path between vertices/atoms *u* and *v* of *G*. So, from Figure 1, there are distances from one to five, for every vertices  $u, v \in V(G)$ .

In other words,  $\forall u, v \in V(G), \exists d(u, v) \in \{1, 2, 3, 4, 5\}$  and obviously  $|D_1| + |D_2| + |D_3| + |D_4| + |D_5| = 66$ . So, we will have five partitions for proof.

I. If d(u,v)=1, then  $|D_1|=12$  and is equal to  $|E(PAH_1)|$ . So, we have three subsets of it.

*I*-1. For six carbon (C) atoms:  $|D_1(C)| = |\{(u,v)| | u, v \in V(PAH_1), d(u,v) = 1 \& d_u + d_v = 6, d_u \times d_v = 9\}|=6$ . Therefore, we have three terms  $6x^1$ ,  $36x^1$ ,  $54x^1$  of the Hosoya, Schultz and Modified Schultz polynomials, respectively.

*I*-2. For six hydrogen (H) atoms:  $|D_1(H)| = |\{(u,v)| | u,v \in V(PAH_1), d(u,v)=1 \& d_u+d_v=4, d_u\times d_v=3\}|=6$ . So, we have three sentences  $6x^l$ ,  $24x^l$  and  $18x^l$  of the Hosoya, Schultz and Modified Schultz polynomials, respectively.

In general, we have three terms  $12x^{l}$ ,  $60x^{l}$  and  $72x^{l}$  for of the above polynomials, respectively.

*II*. If d(u, v)=2, then  $|D_2|=2\times 6_{CH}+1\times 6_{CC}=18$  and similarly, we have

*II*-1. For carbon atoms:  $|D_2(C)| = |\{(u,v)| | u, v \in V(PAH_1), d(u,v)=2 \& d_u + d_v = 6, d_u \times d_v = 9\}|=12$ . Hence, we have three terms  $12x^2$ ,  $72x^2$ ,  $108x^2$  of the Hosoya, Schultz and Modified Schultz polynomials, respectively.

*II-2.* For hydrogen atoms:  $|D_2(H)| = |\{(u,v)| u, v \in V(PAH_1), d(u,v)=2 \& d_u + d_v = 4, d_u \times d_v = 3\}|=6$ . Hence, we have three terms  $6x^2$ ,  $24x^2$ ,  $18x^2$  of these above polynomials, respectively.

So generally, the second term of the Hosoya, Schultz and Modified Schultz polynomials are  $18x^2$ ,  $96x^2$  and  $180x^2$ , respectively.

III. If d(u, v)=3, then  $|D_3|=1\times 6_{HH}+2\times 6_{CH}+3_{CC}=21$  and so, we have three subsets of it.

*III*-1. For carbon atoms:  $|D_3(C)| = |\{(u,v)|u,v \in V(PAH_1), d(u,v)=3 \& d_u+d_v=6, d_u \times d_v=9\}|=3$  and we have three terms  $3x^3$ ,  $18x^3$ ,  $27x^3$  of the Hosoya, Schultz and Modified Schultz polynomials, respectively.

*III-2.* For hydrogen (H) atoms:  $|D_3(H)| = |\{(u,v)|u,v \in V(PAH_1), d(u,v)=3 \& d_u+d_v=2, d_u\times d_v=1\}|=6$ . So, we have three sentences  $6x^3$ ,  $12x^3$  and  $6x^3$  of the Hosoya, Schultz and Modified Schultz polynomials, respectively.

*III*-3. Finally for twelve paths between carbon and hydrogen atoms as distance 3, we have three terms  $12x^3$ ,  $48x^3$ ,  $36x^3$  for the Hosoya polynomial, Schultz polynomial and Modified Schultz polynomial, respectively.

And generally, the thired sentence of the above polynomials are equale to  $21x^3$ ,  $78x^3$  and  $69x^3$ , respectively.

*IV*- If d(u,v)=4, then  $|D_4|=1\times 6_{HH}+6_{CH}=9$  and we have two subsets of  $D_4$  as

*IV*-1. For hydrogen (H) atoms:  $|D_4(H)| = |\{(u,v)|u,v \in V(PAH_1), d(u,v)=4 \& d_u+d_v=2, d_u\times d_v=1\}|=6$ . Thus  $6x^4$ ,  $12x^4$  and  $6x^4$  are in the Hosoya, Schultz and Modified Schultz polynomials, respectively.

*IV-2.* Similarly, for carbon atoms:  $|D_4(C)| = |\{(u,v)|u,v \in V(PAH_1), d(u,v)=4 \& d_u+d_v=4, d_u\times d_v=3\}|=6$  and we have three terms  $6x^4$ ,  $24x^4$  and  $18x^4$ , respectively.

Generally, we will have tow sentences  $36x^4$  and  $24x^4$  of the Schultz and Modified Schultz polynomials and  $12x^4$  of the Hosoya polynomial.

*V*- If d(u,v)=5, then  $|D_5|=|D_5(H)|=|\{(u,v)|u,v \in V(PAH_1), d(u,v)=5 \& d_u+d_v=2, d_u\times d_v=1\}|=3$  and obviusly we have three terms  $3x^5$ ,  $6x^5$  and  $3x^5$  for the Hosoya, Schultz and Modified Schultz polynomials of  $PAH_1$ , respectively.

Now, by enumerate all distinct shortest path between vertices/atoms u, v of Benzene PAH<sub>1</sub>, its Hosoya polynomial is equal to:

$$H(PAH_1,x) = 12x^1 + 18x^2 + 21x^3 + 12x^4 + 3x^5$$

And the Wiener index of *Benzene*  $PAH_1$  is as follow:

$$W(PAH_1) = \frac{\partial H(PAH_1, x)}{\partial x} = 12 \times 1 + 18 \times 2 + 21 \times 3 + 12 \times 4 + 3 \times 5 = 164.$$

The Schultz polynomial of  $PAH_1$  is equal to:

$$SC(PAH_1,x) = 60x^1 + 96x^2 + 78x^3 + 36x^4 + 6x^5$$

And the Schultz index of  $PAH_1$  is equal to:

$$SC(PAH_{I}) = \frac{\partial SC(PAH_{1}, x)}{\partial x} = 60 \times 1 + 96 \times 2 + 78 \times 3 + 36 \times 4 + 6 \times 5 = 660$$

Finally, the Modified Schultz polynomial and Modified Schultz index of *PAH*<sub>1</sub> are equale to:

$$SC^{*}(PAH_{1},x) = 72x^{1} + 180x^{2} + 69x^{3} + 24x^{4} + 3x^{5}$$

And also,

$$SC^{*}(PAH_{1}) = \frac{\partial SC^{*}(PAH_{1}, x)}{\partial x} = 72 \times 1 + 180 \times 2 + 69 \times 3 + 24 \times 4 + 3 \times 5 = 750$$

Here, we complete the proof of Theorem 1.  $\Box$ 

## Acknowledgement

The author is thankful to *Prof. M.V. Diudea* and *Dr. M.P. Vlad* from Faculty of Chemistry and Chemical Engineering Babes-Bolyai University and Faculty of Economic Sciences of Dimitrie Cantemir University (Romania) and *Dr. M. Alaeiyan* and *S.H. Hosseini* of Department of Mathematics, Iran University of Science and Technology (IUST) for their helpful comments and suggestions.

## References

[1] H. Wiener, Structural determination of paraffin boiling points, J. Amer. Chem. Soc. 69 (1947), 17–20.

[2] H. Hosoya. On some counting polynomials in chemistry. Discrete Appl. Math. 19, (1988), 239-257.

[3] H.P. Schultz, Topological organic chemistry 1. Graph theory and topologicalindices of alkanes. J. Chem. Inf. Comput. Sci. 29 (1989), 227–228.

[4] S. Klavžar and I. Gutman. Wiener number of vertex-weighted graphs and a chemical application. Disc. Appl. Math. 80 (1997), 73–81.

[5] Y. Alizadeh, A. Iranmanesh and S. Mirzaie, Computing Schultz Polynomial, Schultz Index of C60 Fullerene By Gap Program. Digest. J. Nanomater. Bios. 4(1), (2009), 7-10.

[6] M. Eliasi and B. Taeri, Schultz Polynomials of Composite Graphs, Appl. Anal. Discrete Math. 2, (2008), 285-296.

[7] O. Halakoo, O. Khormali and A. Mahmiani, Bounds For Schultz Index of Pentachains, Digest. J. Nanomater. Bios. 4(4), (2009), 687-691.

[8] A. Heydari. On The Modified Schultz Index of C4C8(S) Nanotubes. Digest. J. Nanomater. Bios. 5(1), (2010), 51-56.

[9] A. Iranmanesh and Y. Alizadeh, Computing Hyper-Wiener and Schultz Indices of TUZ6[p.q] Nanotube By Gap Program, Digest. J. Nanomater. Bios. 4(1). (2009), 607-611.

[10] A. Iranmanesh and Y. Alizadeh, Computing Szeged and Schultz Indices of HAC5C7C9[p,q] Nanotube By Gap Program, Digest. J. Nanomater. Bios. 4(1), (2009), 67-72.

[11] S. Klavžar and I. Gutman, Bounds for The Schultz Molecular Topological Index of benzenoid Systems in Terms of Wiener Index. J. Chem. Inf. Comput. Sci. 37(4), (1997), 741-744.

[12] S. Klavžar and I. Gutman, A Comparison of The Schultz Molecular Topological Index with The Wiener Index. J. Chem. Inf. Comput. Sci. 36(5), (1996), 1001–1003.

[13] M.R.Farahani. On the Schultz and Modified Schultz Polynomials of Some Harary Graphs. International Journal of Applications of Discrete Mathematics. 1(1), (2013), 1-8.

[14] M.R. Farahani, M.P. Vlad. On the Schultz, Modified Schultz and Hosoya polynomials and Derived Indices of Capra-designed planar Benzenoids. Studia Universitatis Babes-Bolyai Chemia. 4, (2012) 55-63.

[15] M.R. Farahani, On the Schultz polynomial, Modified Schultz polynomial, Hosoya polynomial and Wiener index of circumcoronene series of benzenoid. Journal of Applied Mathematics & Informatics. 31(5-6) in press, (2013).

[16] C.W. Bauschlicher, Jr, E.L.O. Bakes, Chem. Phys. 2000, 262, 285-291.

[17] A. J. Berresheim, M.M. ller, K.M. llen, Chem. Rev. 1999, 99, 1747-1785.

[18] A. M. Craats, J.M. Warman, K.M. Ilen, Y. Geerts, J. D. Brand, Adv. Mater. 1998, 10, 36-38.

[19] F. Dietz, N. Tyutyulkov, G.Madjarova, K.M. llen, J. Phys. Chem. B 2000, 104, 1746-1761.

[20] F. Dtz, J. D. Brand, S. Ito, L. Ghergel, K.M. Ilen, J. Am. Chem. Soc. 2000, 122, 7707-7717.

[21] M.R. Farahani, Some Connectivity Indices of Polycyclic Aromatic Hydrocarbons (PAHs). Submitted for publish, (2013).

[22] M.R. Farahani. Zagreb Indices and Zagreb Polynomials of Polycyclic Aromatic Hydrocarbons. J.Chem. Acta. 2, (2013), 70-72.

[23] S.E. Huber, A. Mauracher and M.Probst. Permeation of Low-Z Atoms through Carbon Sheets: Density Functional Theory Study on Energy Barriers and Deformation Effects. Chem. Eur. J. 2003, 9, 2974-2981.

[24] K. Jug, T. Bredow, Models for the treatment of crystalline solids and surfaces, J. Computational Chemistry, 25 (2004) 1551-1567.

[25] A. Soncini, E. Steiner, P.W. Fowler, R.W. A. Havenith and L.W. Jenneskens. Perimeter Effects on Ring Currents in Polycyclic Aromatic Hydrocarbons: Circumcoronene and Two Hexabenzocoronenes. Manuscript.

[26] S.E. Stein, R.L. Brown, J. Am. Chem. Soc. 1987, 109, 3721-372.

[27] M. Wagner, K.M. llen, Carbon 1998, 36, 833-83.

[28] U.E. Wiersum, L.W. Jenneskens in Gas Phase Reactions in Organic Synthesis, (Ed.: Y. Valle. e), Gordon and Breach Science Publishers, Amsterdam, The Netherlands, 1997, 143–194.

[29] K.Yoshimura, L. Przybilla, S. Ito, J.D. Brand, M. Wehmeir, H.J. Rder, K.M. llen, Macromol. Chem. Phys. 2001, 202, 215-222.

[30] J. Brunvoll, B.N. Cyvin and S.J. Cyvin. Enumeration and Classification of Benzenoid Hydrocarbons. Symmetry and Regular Hexagonal Benzenoids. J. Chem. If. Comput. Sci. 27, 171-177. (1987).

[31] J.R. Dias. From benzenoid hydrocarbons to fullerene carbons. MATCH Commun. Math. Comput. Chem. 4, 57-85. (1996).

[32] M.V. Diudea, M. Ştefu, P.E. John, and A. Graovac. Generalized operations on maps, Croat. Chem. Acta, 2006, 79, 355-362.

[33] M.V. Diudea. Nano porous carbon allotropes by septupling map operations. J. Chem. Inf. Model, 2005, 45, 1002-1009.

[34] A. Dress and G. Brinkmann. Phantasmagorical fulleroids, MATCH Commun. Math. Comput. Chem. 33, 87-100. (1996).

[35] E. Estrada, L. Torres, L. Rodriguez and I. Gutman. Indian J. Chem. 37, 849 (1998).

[36] M.R. Farahani. Zagreb index, Zagreb Polynomial of Circumcoronene Series of Benzenoid. Adv. Mater. Corrosion. 1, (2012), In press.

[37] M. Goldberg. A class of multi-symmetric polyhedra. Tohoku Math. J. 1937. 43, 104-108.

[38] I. Gutman and S. Klavžar, M. Petkovsek and P. Zigert. On Hosoya polynomials of benzenoid graphs, MATCH Commun. Math. Comput. Chem. 43, 49–66 (2001).

[39] I. Gutman and S. Klavžar. A method for calculationg Wiener numbers of benzenoid hydrocarbons and phenylenes. ACH Mo dels Chem. 133, 389-399. (1996).

[40] S. Klavžar. A Bird's Eye View of The Cut Method And A Survey of Its Applications In Chemical Graph Theory. MATCH Commun. Math. Comput. Chem. 60, 255-274. (2008).

[41] S. Klavžar, I. Gutman and B. Mohar. Labeling of Benzenoid Systems which Reflects the Vertex-Distance Relations. J. Chem. Int Comput. Sci. 35, 590-593. (1995).

[42] K. Salem, S. Klavžar and I. Gutman. On the role of hypercubes in the resonance graphs of benzenoid graphs. Discrete Mathematics. 13(8) 306, (2003).

[43] V. Chepoi and S. Klavžar. Distances in benzenoid systems: Further developments. Discrete Mathematics. 192, 27-39. (1998).